

tion of the adducts (5).⁴ X-Ray and n.m.r. studies have already shown that, in the presence of certain ligands, the structure of (3) is more closely represented as a σ - π -allylnickel species (4).⁵ From the present results, it appears that reaction of the aldehyde occurs preferentially at the σ -allylnickel bond and amines and active methylene compounds react at the π -allyl sites.

Reactions with isoprene point to a similar situation on treatment with the complex (6). In the presence of NiCl₂ (3.3 mmol), PPh₃ (6.6 mmol), and NaBH₄ (13.3 mmol) in ethanol, isoprene (0.13 mol) and diethyl malonate (0.043 mol) gave, after 24 h at 20 °C, a 64% conversion into a mixture containing (8) (14%), (9) (17%), (10) (39%), (11) (10%), and (12) (15%). Again, physical data indicate that, in the presence of ligands, the bis- π -allylic complex (6) displays the σ - π -structure (7).⁶ Whilst some 1:1 adducts, and a product derived from tail-tail isoprene dimerisation, are also obtained, the predominant adduct is formed by reaction of the nucleophilic reagent at the π -allyl site of (7).

A solution of isoprene (200 mmol) and cyclododecatriene-triphenylphosphinenickel,⁷ derived from nickel acetylacetonate (19.4 mmol), in ether, was stirred overnight at

0 °C, cooled to -30 °C, and acetaldehyde (88 mmol) was added. The solution was stirred for a further 6 h whilst allowing it to warm to room temperature. Addition of saturated aqueous NaCN and extraction with ether gave both 2:1 (40%) and 1:1 (18%) isoprene-aldehyde adducts.‡ The former consisted of mainly (13) (>90%, 13a:13b, 1:1) and the major product of the latter was (14) (65%) together with (15) (35%). The reaction of aldehyde, therefore, indicates a preference for attack at the σ -allyl group of (7).

It is clear that the predominant products from reaction of amines and active methylene compounds with (4) and (7) are derived by attack at the π -allyl site whilst aldehydes undergo preferential reaction at the σ -allyl group of the nickel complexes. The specificity observed in the present reactions might be generally applicable to other nucleophilic and electrophilic reagents. Further studies will indicate if the selectivity is related to the reactivity of the electrophile and the electronic properties of the organophosphorous ligands.

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